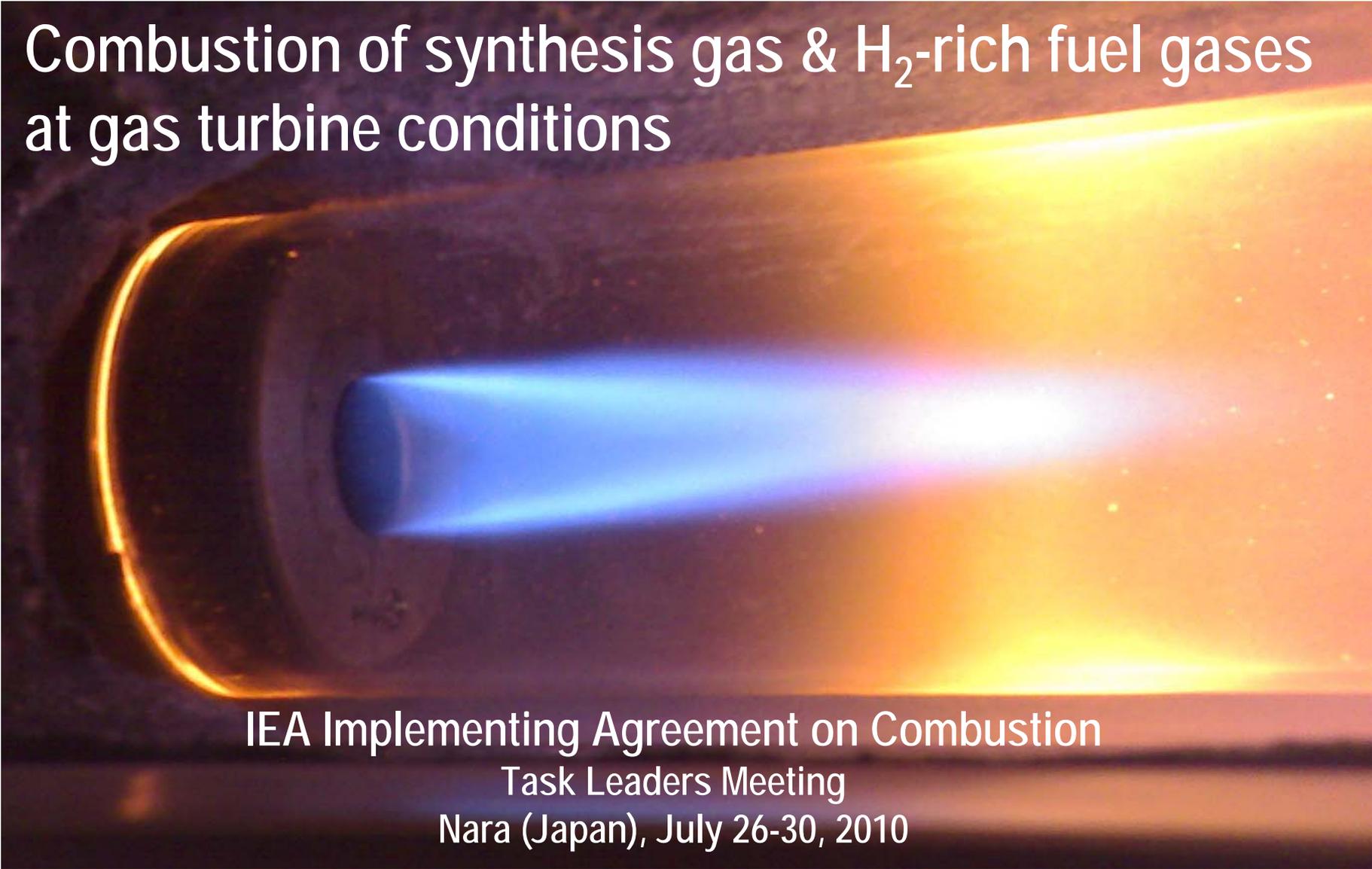
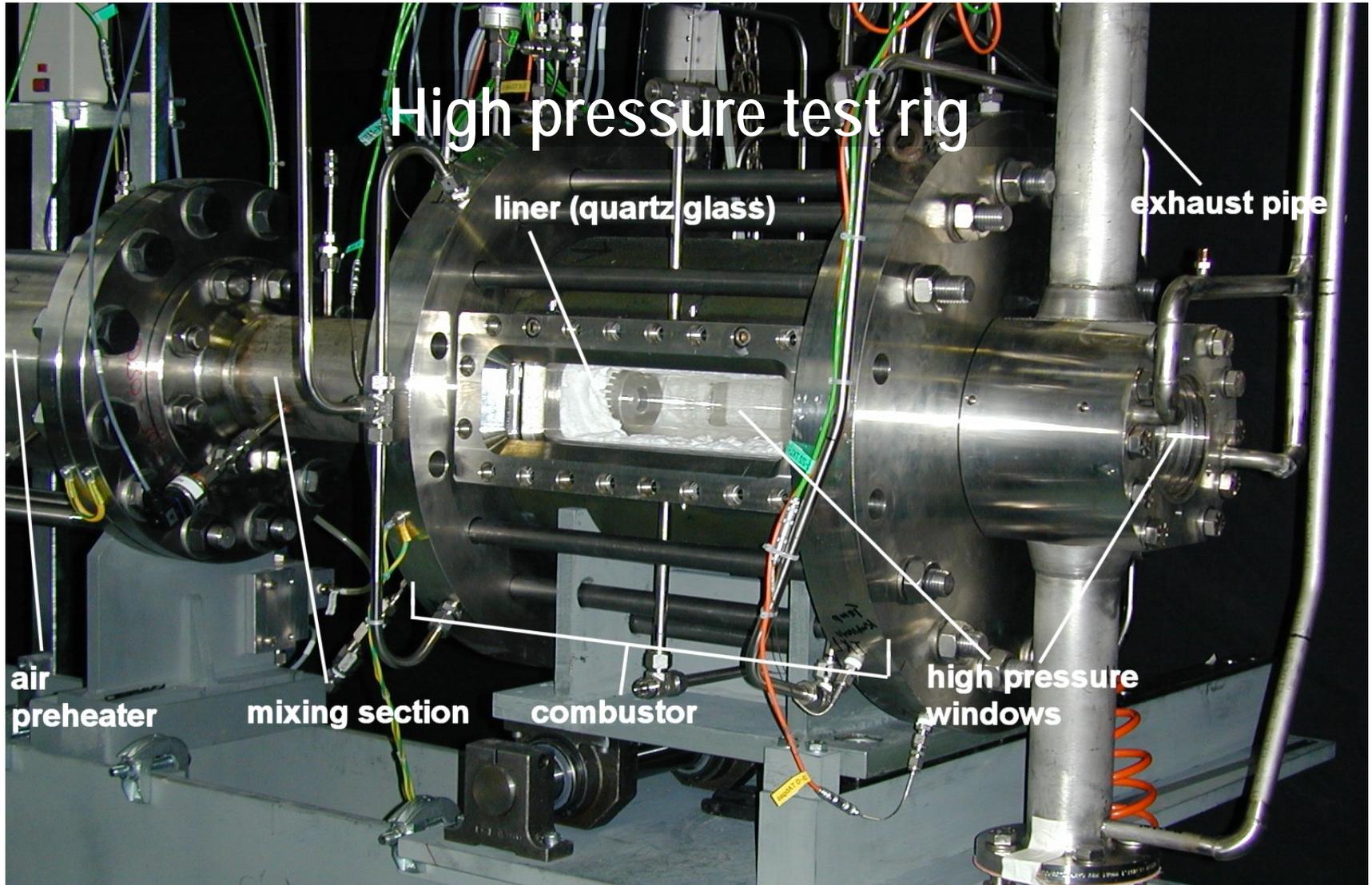


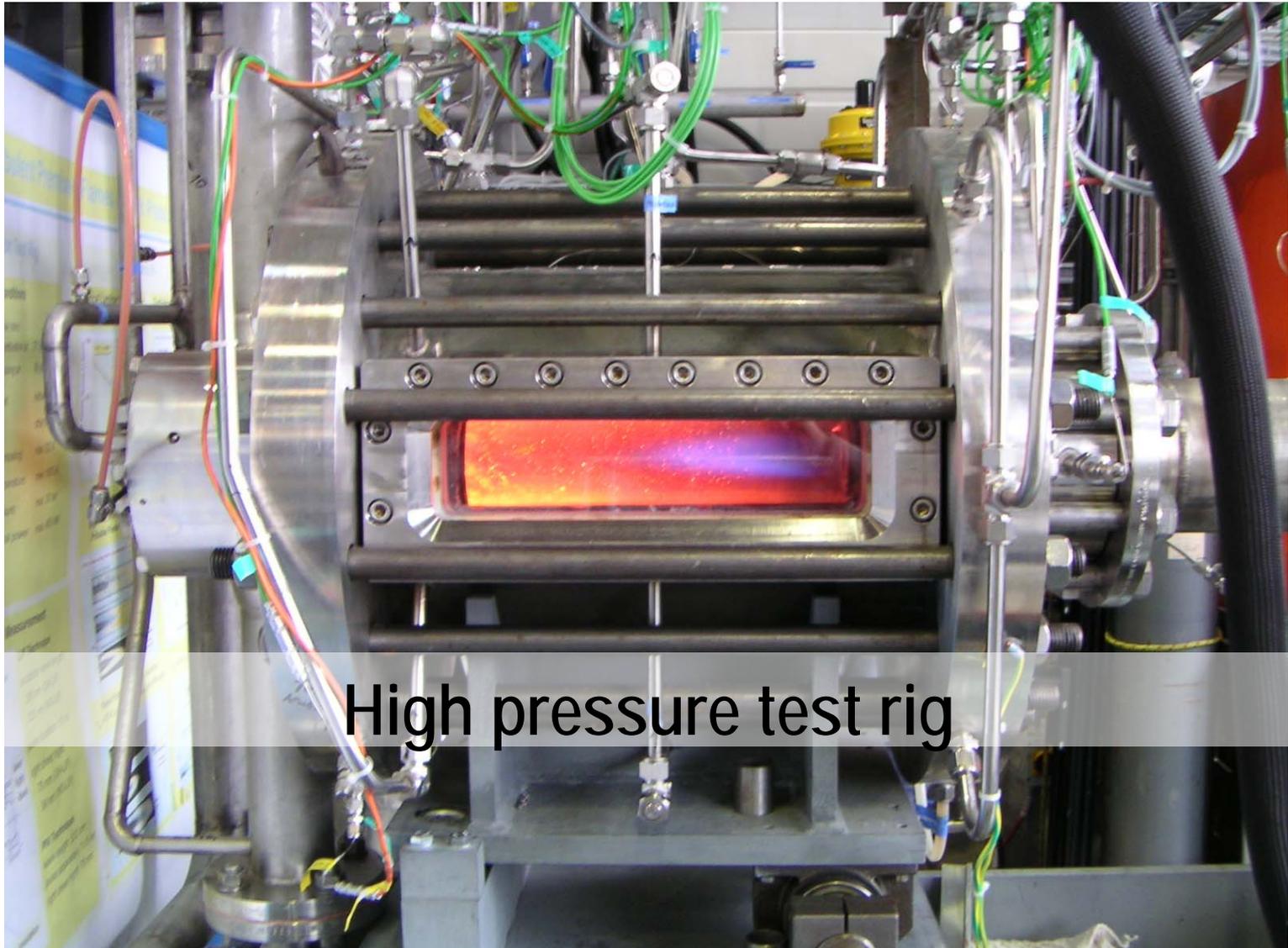
Combustion of synthesis gas & H₂-rich fuel gases at gas turbine conditions

A photograph of a gas turbine combustion chamber. A bright blue flame is visible at the inlet, transitioning to a bright yellow and orange flame as it moves through the chamber. The chamber is dark and metallic, with a curved, cylindrical shape.

IEA Implementing Agreement on Combustion

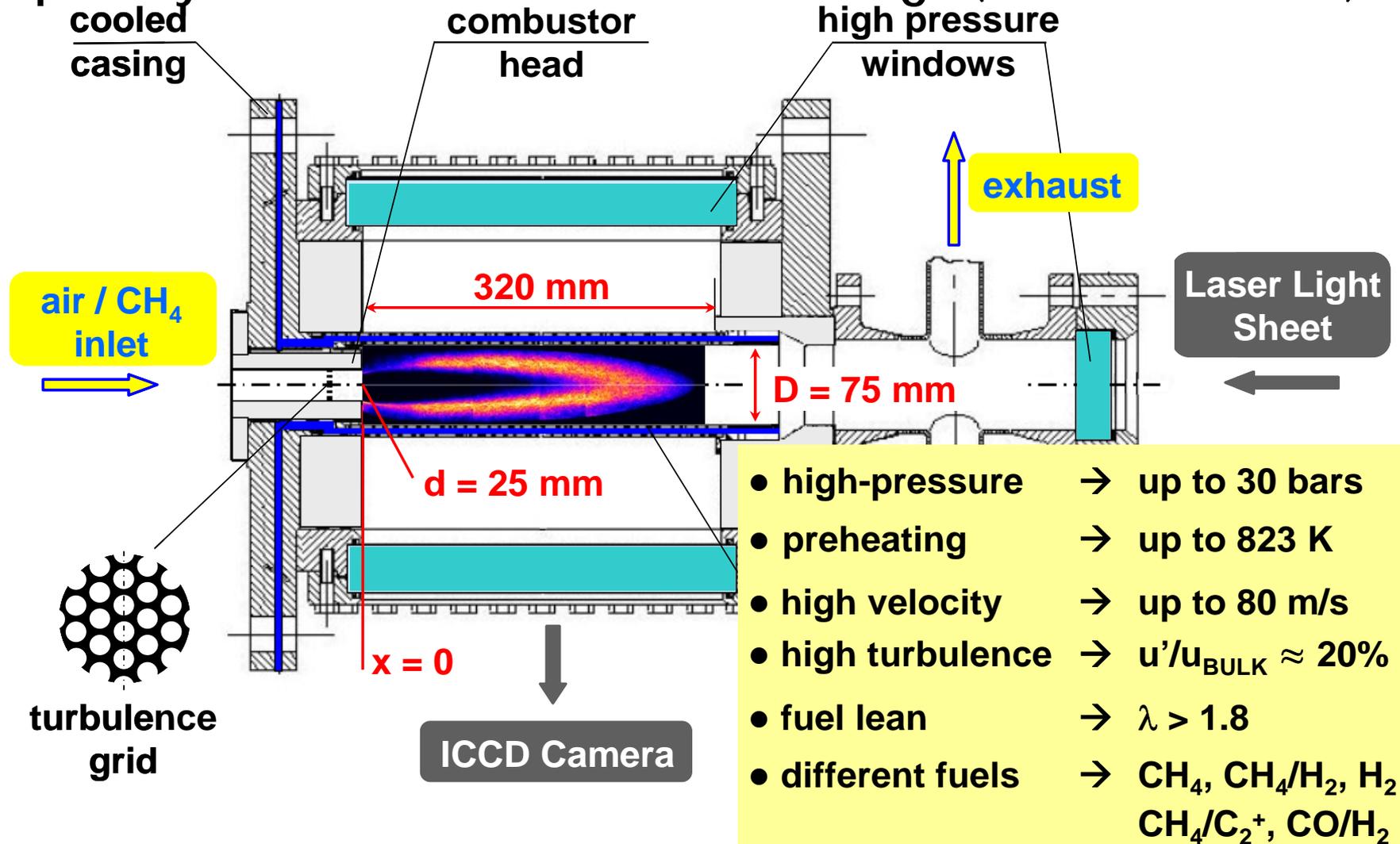
Task Leaders Meeting
Nara (Japan), July 26-30, 2010



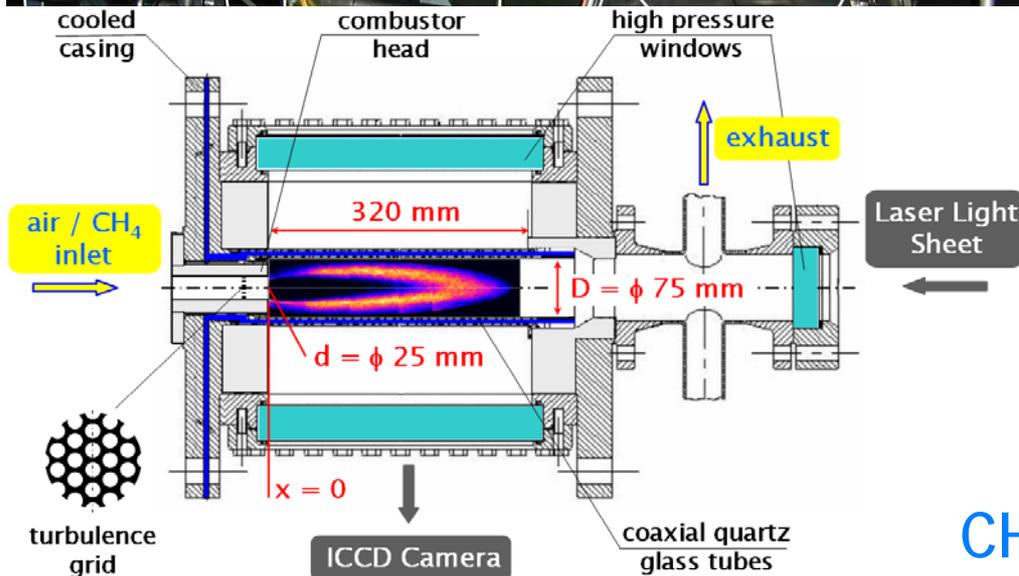
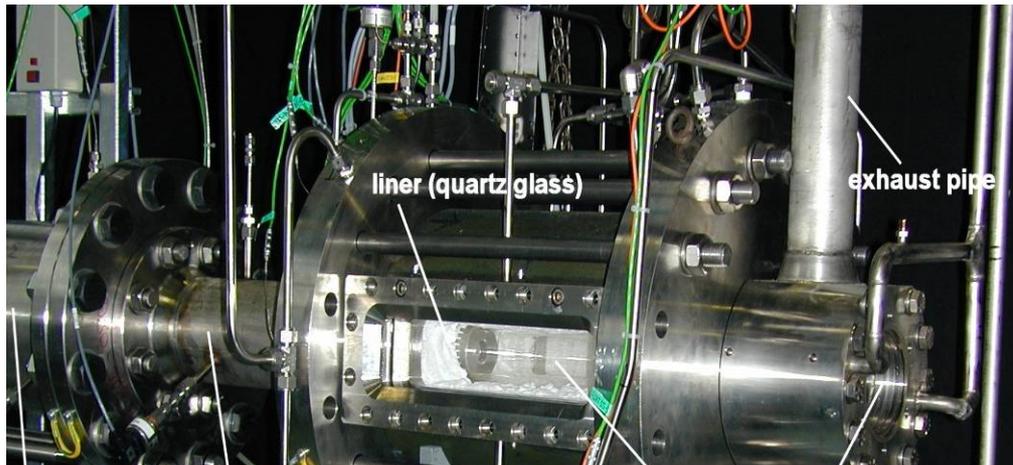


High pressure test rig

Optically accessible combustion test rig (LIF, CL, Raman, ...)



Turbulent Lean Premix Combustion



Objectives

lean premixed combustion technology for gas turbines

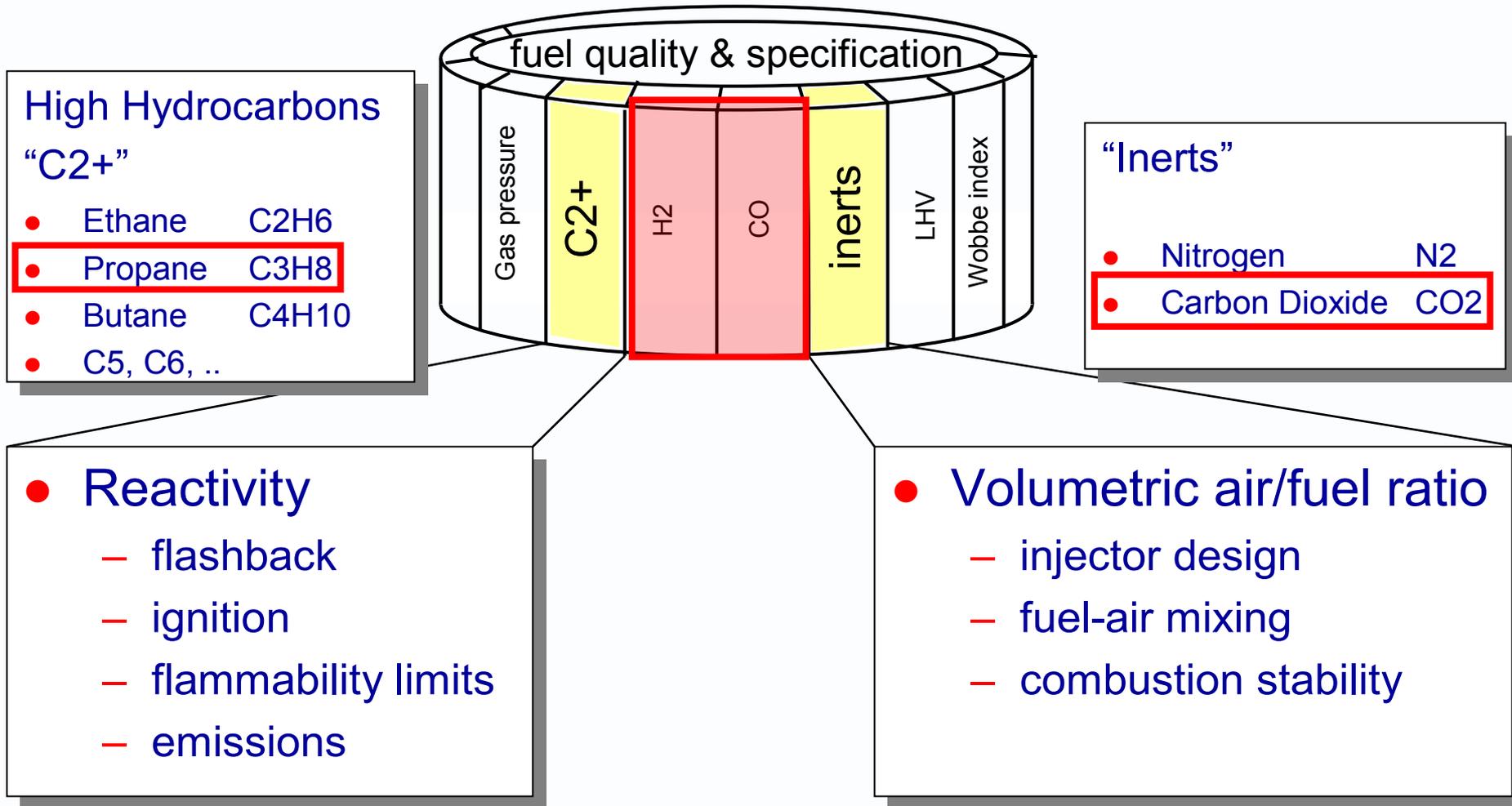
- **enhancing flame stability**
(extension of lean blow-out limit;
prevent flashback;
reduce flame front fluctuations)
- **lower NO_x emissions**
(improved fuel/air mixing)

flame characteristics of lean premixed flames

- **influence of turbulence**
- **broad fuel spectrum**

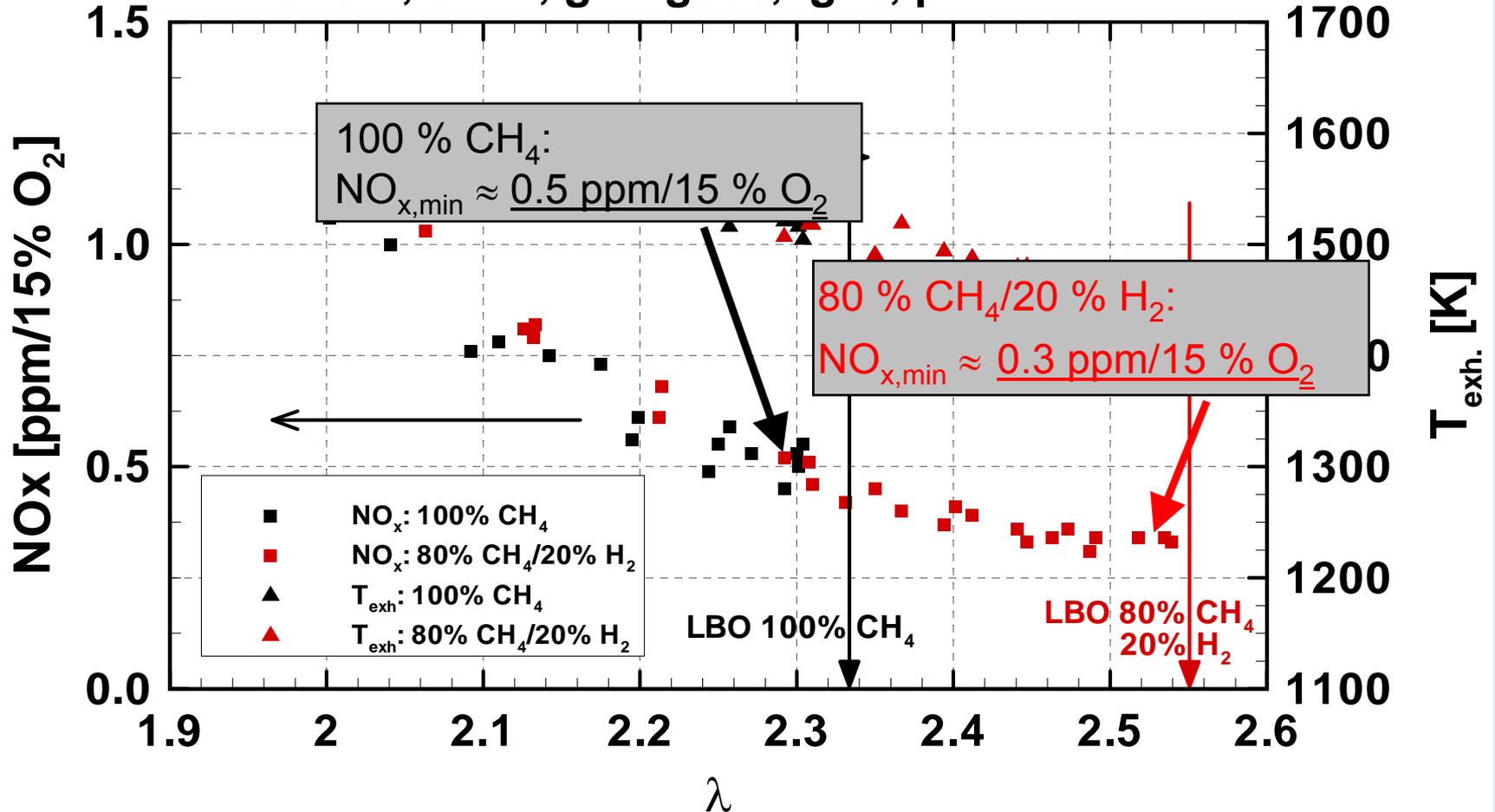
$\text{CH}_4, \text{CH}_4/\text{H}_2, \text{CH}_4/\text{C}_3\text{H}_8, \text{CO}/\text{H}_2, \text{H}_2$

Combustion Impact of Gas Composition

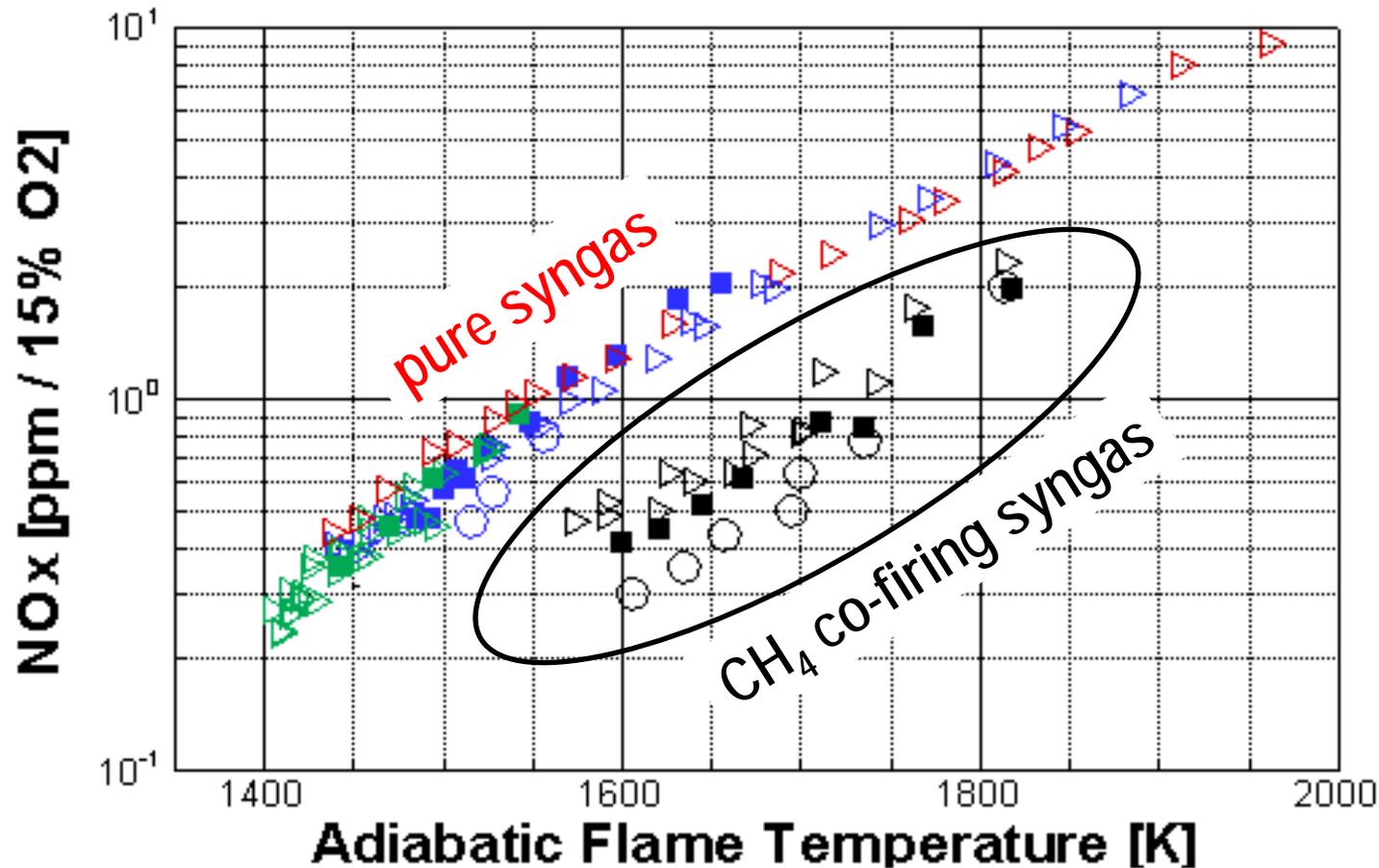


NO_x emissions

673 K, 5 bar, grid g365,xg10, premixed



NO_x emission depend on gas mixture and temperature

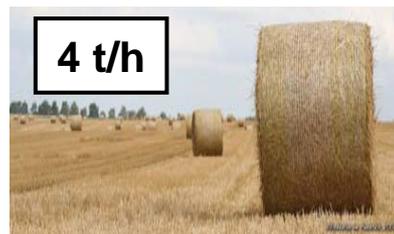


Lower NO_x for co-firing mixture: Different pathway for C_xH_y oxidation

CO₂ MITIGATION VIA CO-FIRING OF BIOMASS

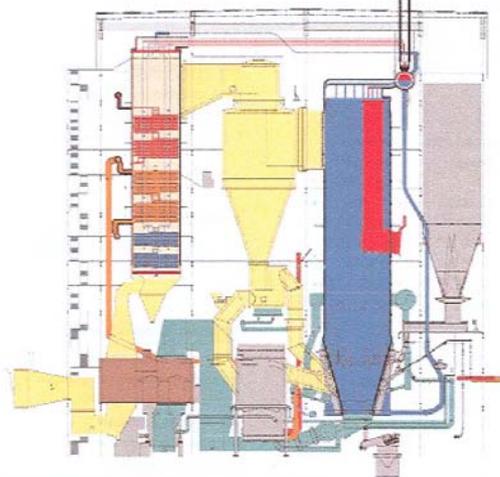


input
for 10 MW_e

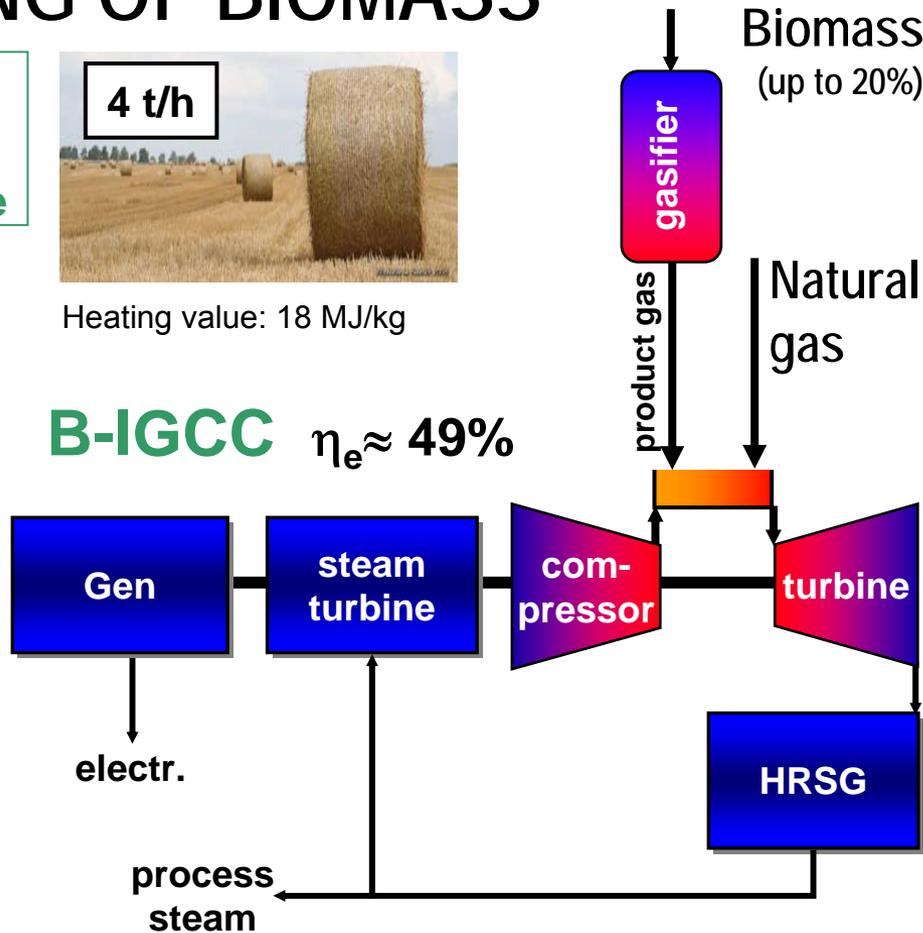


Heating value: 18 MJ/kg

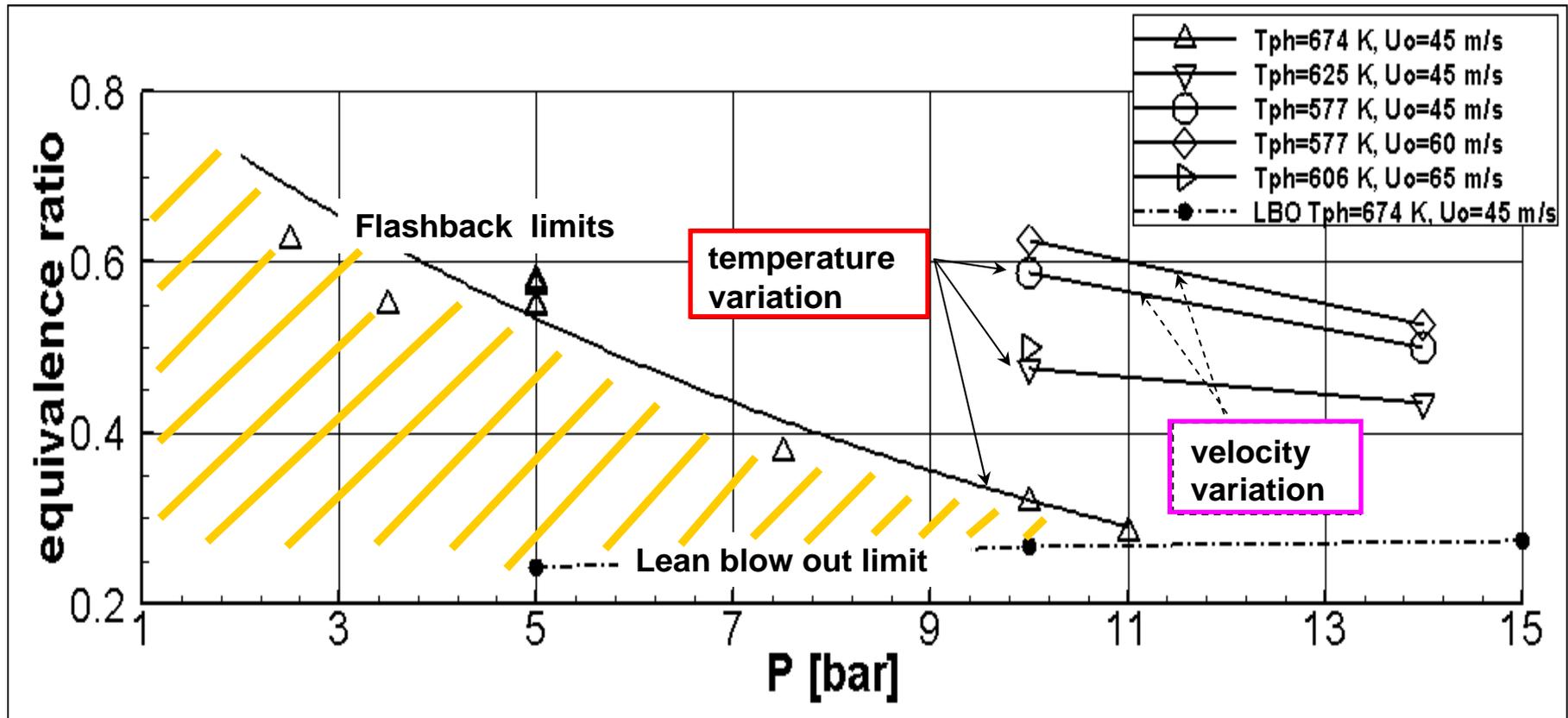
Steam cycle $\eta_e \approx 35\%$



B-IGCC $\eta_e \approx 49\%$



Operational window (syngas: 50% H₂ / 50% CO)



Operational window (syngas: 50% H₂ / 50% CO)

lean blow out

equivalence ratio

lifted flame

stable flame

$$\phi = \left(\frac{F}{A} \right) / \left(\frac{F}{A} \right)_{st}$$

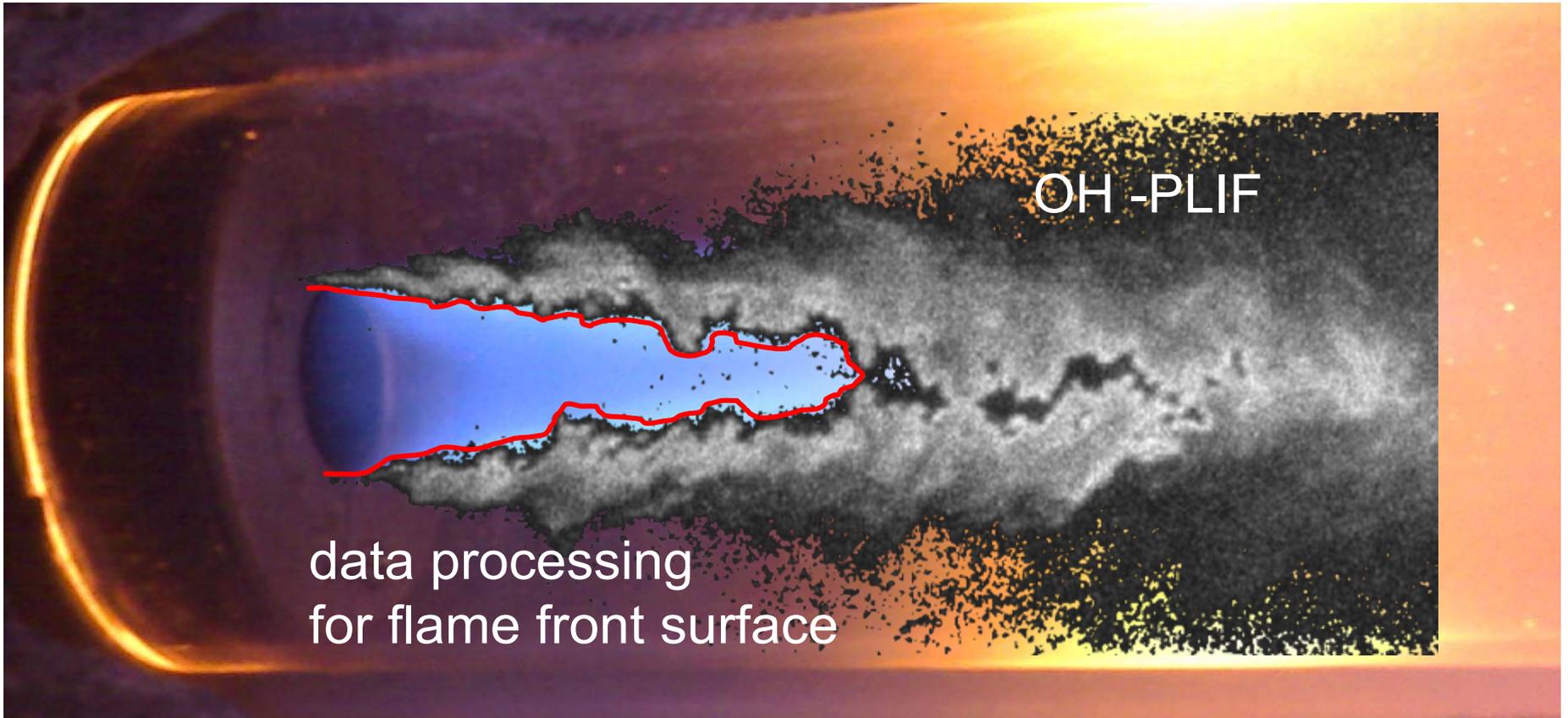
flame stabilized close to sudden expansion

flash back

flame speed overtakes flow velocity

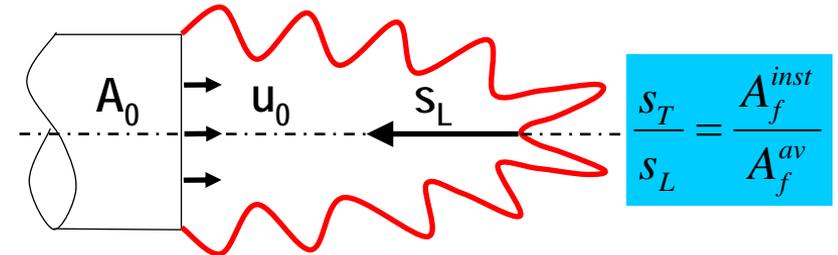
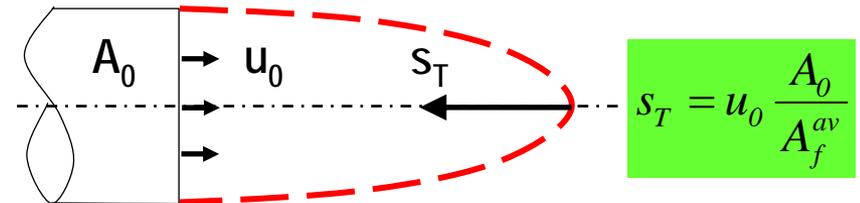
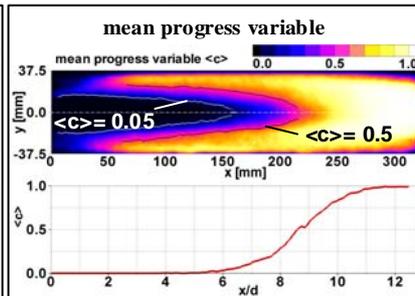
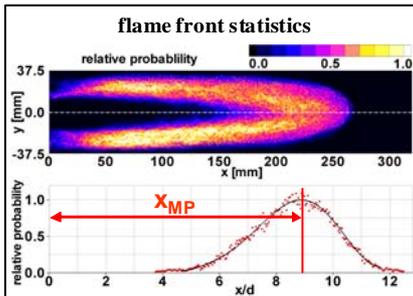
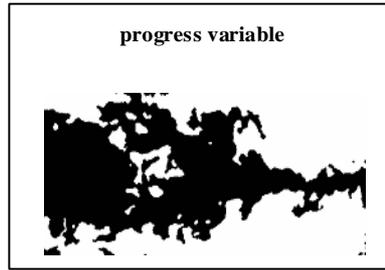
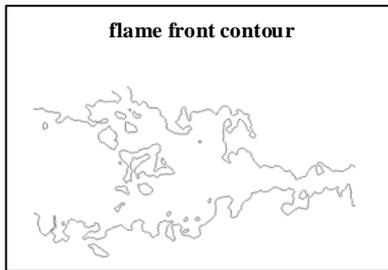
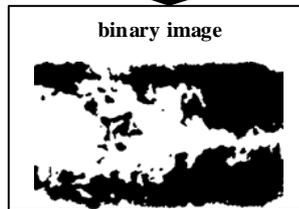
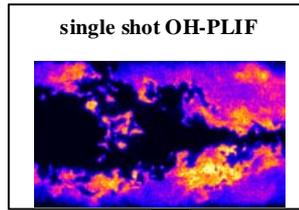
locally and/or globally

Flame Front Detection & Characterization

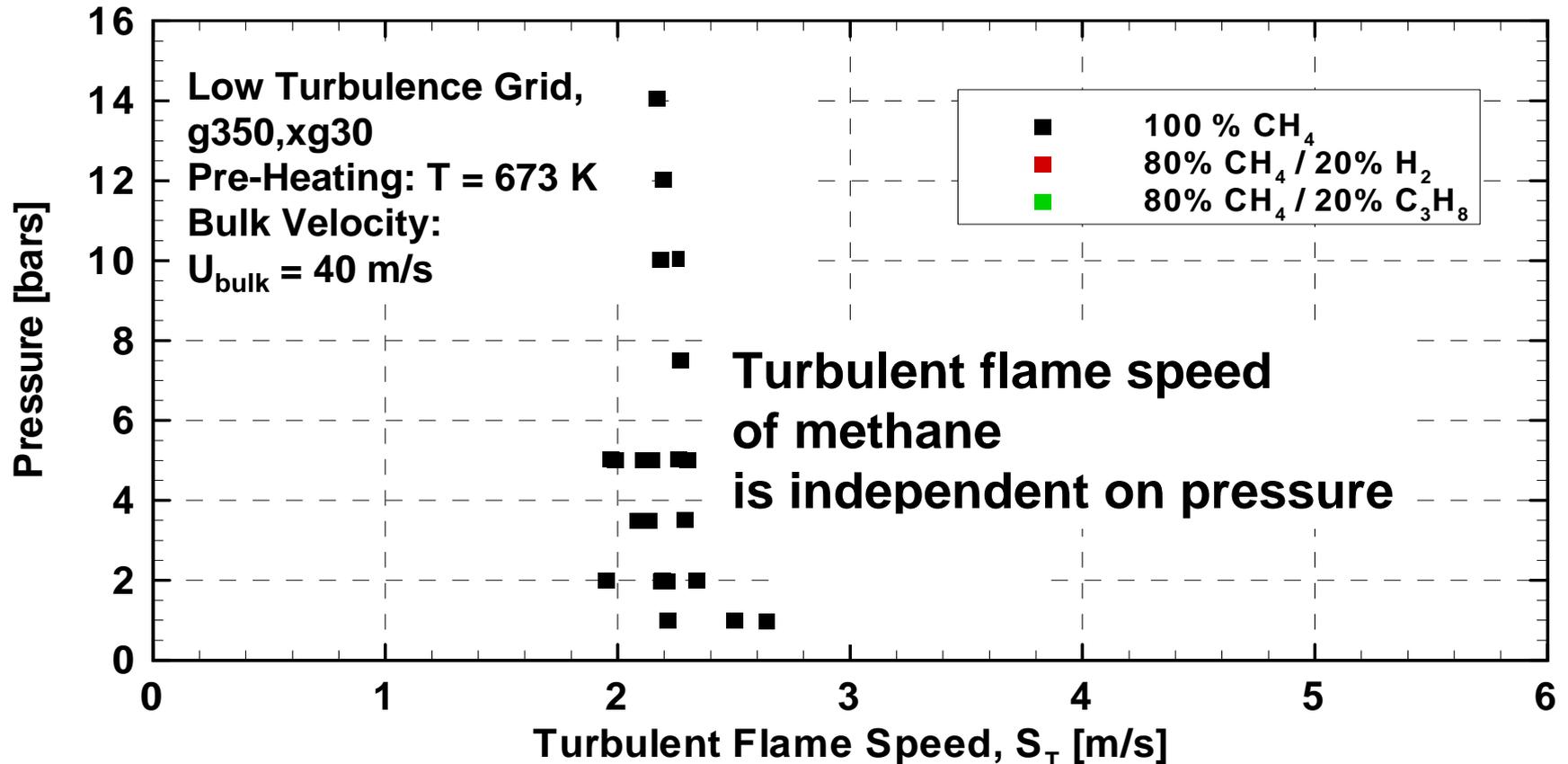


How to determine turbulent flame speed (S_T) data

$$\rho_0 \cdot A_0 \cdot U_0 = \rho_0 \cdot A_f^{av} \cdot S_T = \rho_0 \cdot A_f^{inst} \cdot S_L$$



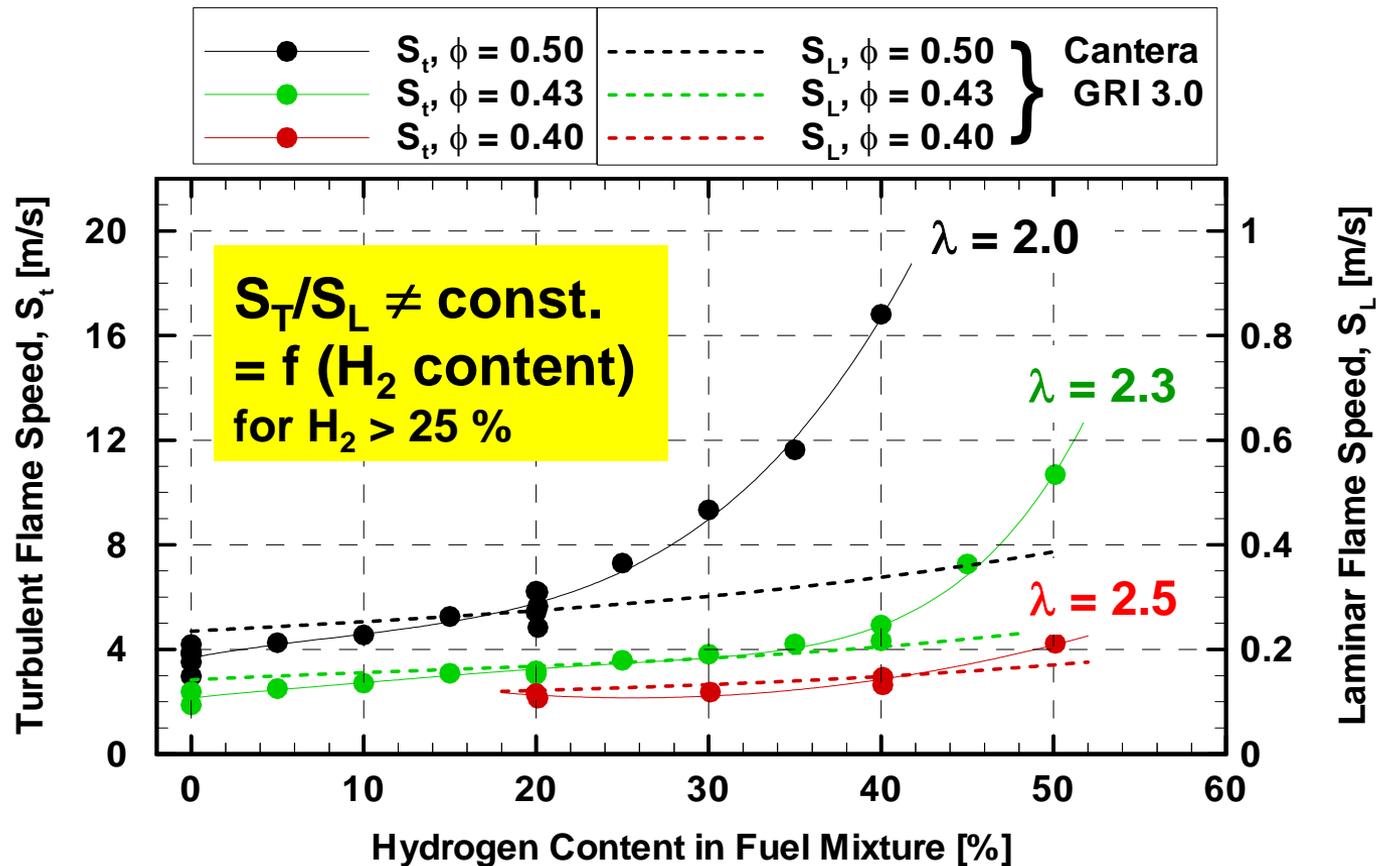
Turbulent Flame Speed



**Turbulent flame speed of methane mixtures (with C₃H₈, H₂)
 is dependent on pressure
 (even though the effect is very small above 10 bar)**

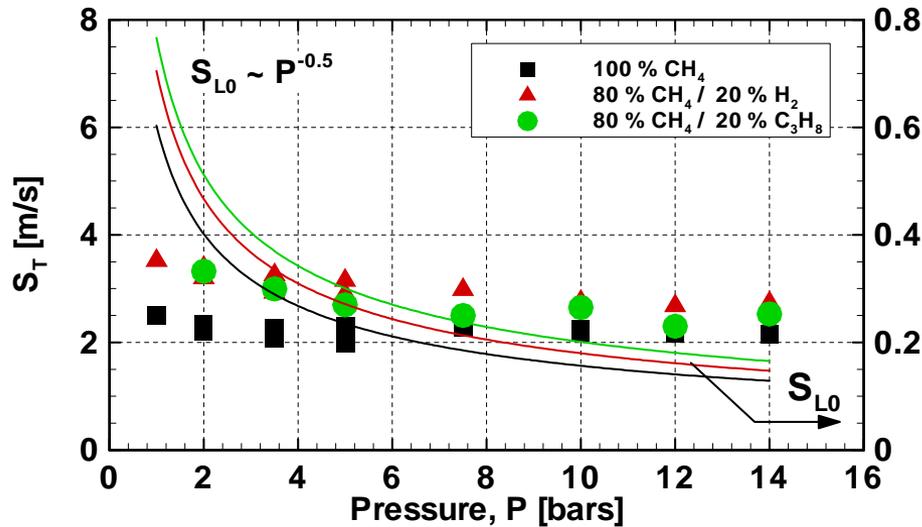
Turbulent Flame Speed

673 K, 5 bar, 40 m/s, grid g365,xg10



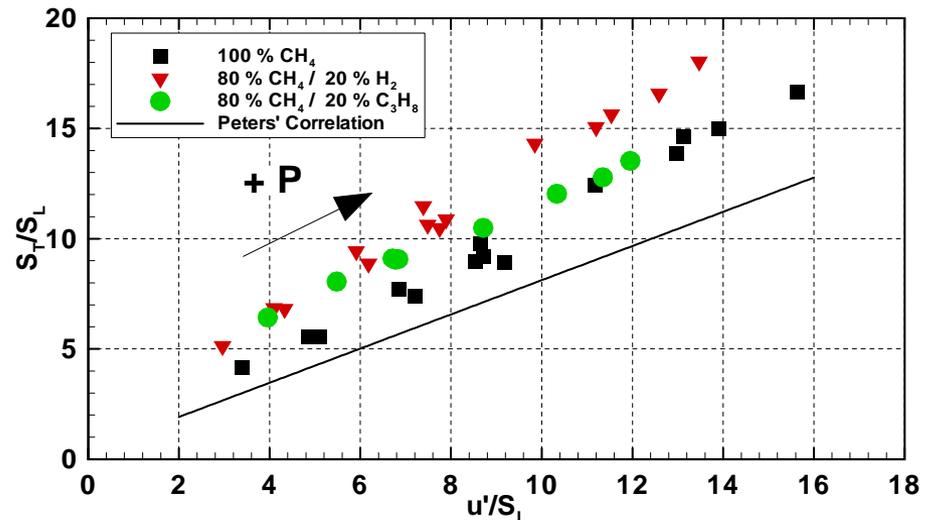
Up to approx. 25 % Vol. H_2 : chemical kinetics dominate (S_T/S_L const.)
 $\text{H}_2 > 25\%$ Vol.: additional effects (preferential diffusion, stretch)

Turbulent Flame Speed: Influence of Pressure

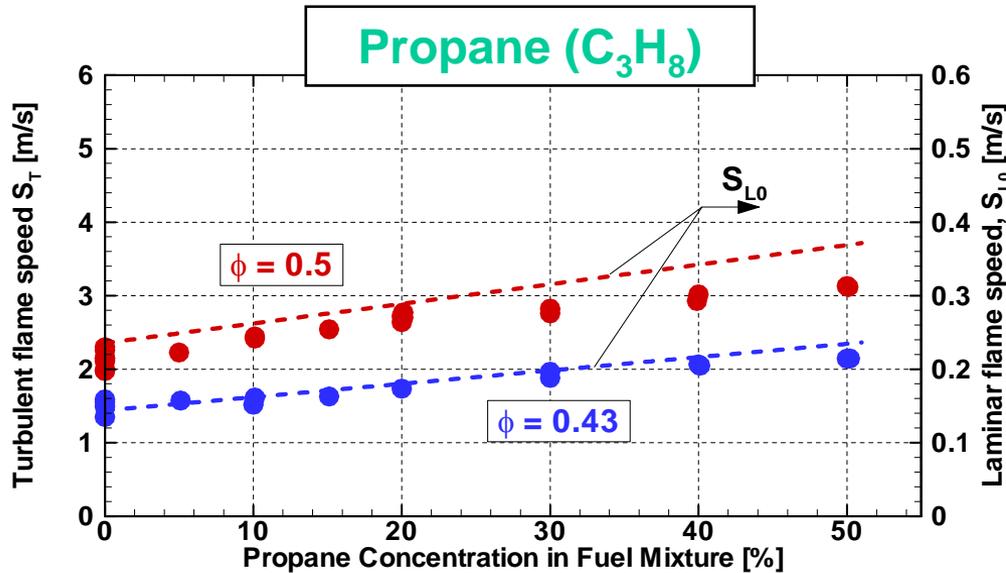


- Laminar flame speed is heavily dependent on pressure
- Turbulent flame speed is nearly independent of pressure for the three mixtures
- The increase in Re_T by increasing the pressure decreases the size of the smallest turbulent eddies

The resulting increase in turbulent flame surface counteracts the decrease of laminar flame speed



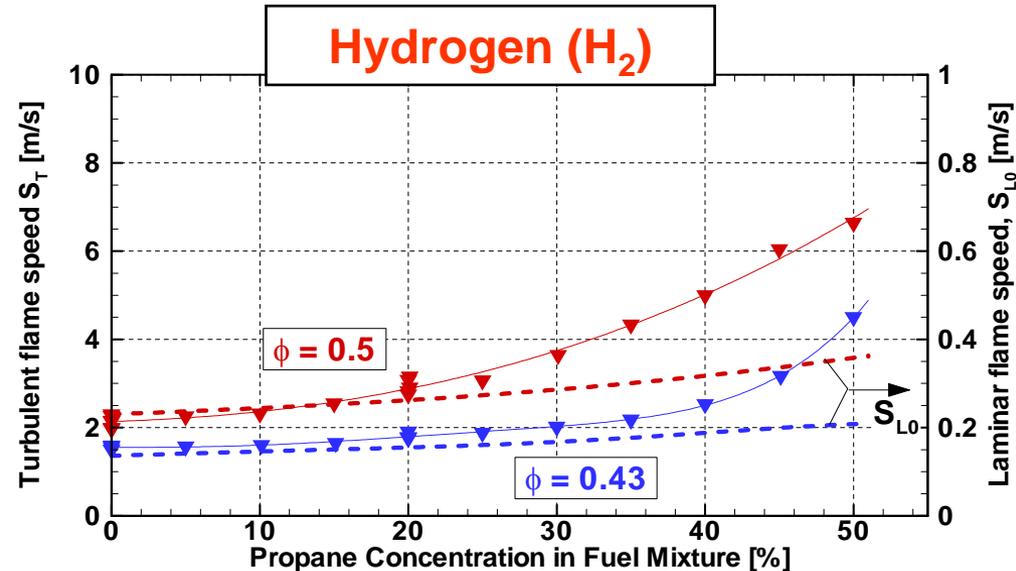
Turbulent Flame Speeds: Influence of Fuel Blend



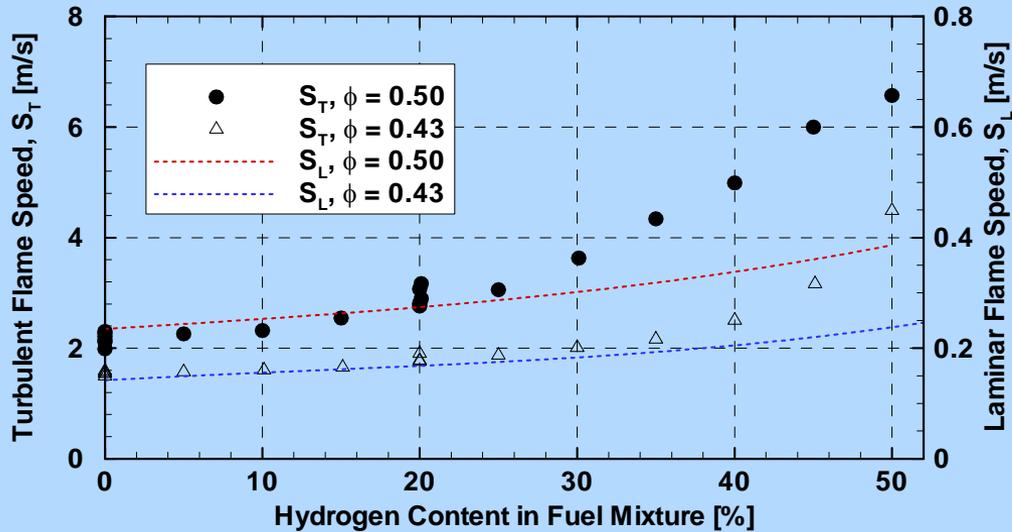
- Effect of C₃H₈ addition:
 - Trend is consistent for both ϕ
 - Turbulent flame speed trend seems to be dominated by chemistry
 - Constant $S_T/S_L \approx 10$ in the measured range (100 % CH₄ – 50 % C₃H₈)

Effect of H₂ addition:

- S_T deviates greatly from S_L trend, especially at higher hydrogen content. i.e. more than just chemical effect
- Effect is less pronounced for leaner mixtures.

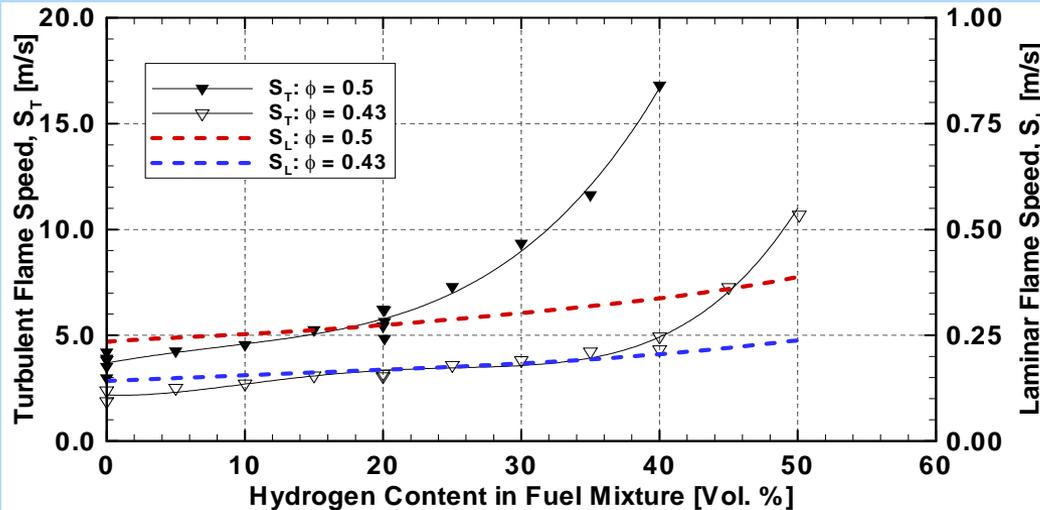


Turbulent Flame Speed



Low Turbulence Grid,
 g350,xg30
 Pre-Heating: $T = 673 \text{ K}$
 Bulk Velocity: $U_{\text{bulk}} = 40 \text{ m/s}$
 Pressure = 5 bar

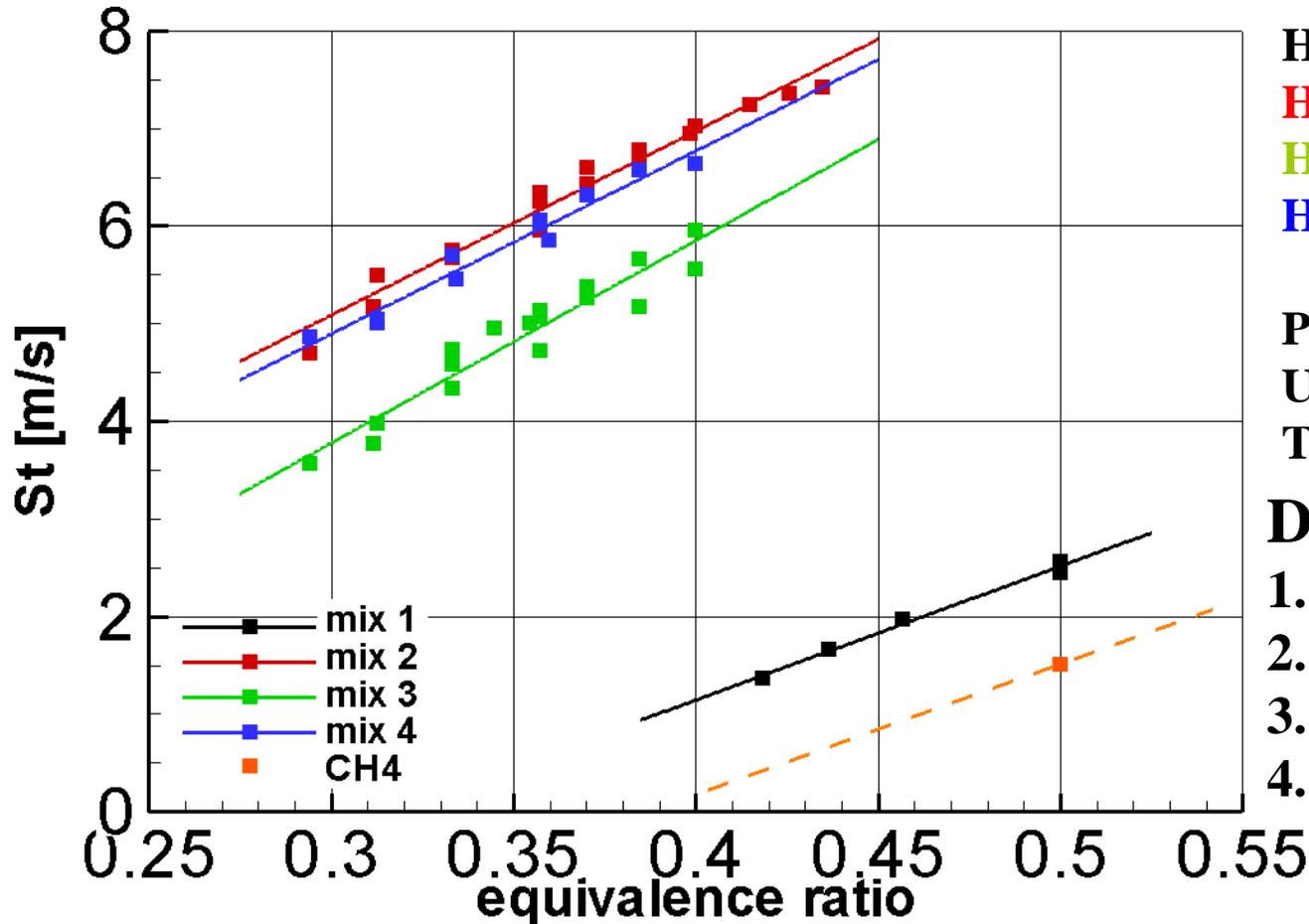
**$S_T/S_L \neq \text{const.}$
 $= f(\text{H}_2 \text{ content})$
 for $\text{H}_2 > 25 \%$**



High Turbulence Grid,
 g365,xg10
 Pre-Heating: $T = 673 \text{ K}$
 Bulk Velocity: $U_{\text{bulk}} = 40 \text{ m/s}$
 Pressure = 5 bar

Turbulent flame speed S_t

- dependency on fuel composition and stoichiometry



H2-CO-CH4 20-20-60 - black

H2-CO 50-50 - red

H2-CO 33-67 - green

H2-CO-N2 40-40-20 - blue

$P = 5$ bar

$U_{\text{bulk}} = 40$ m/s

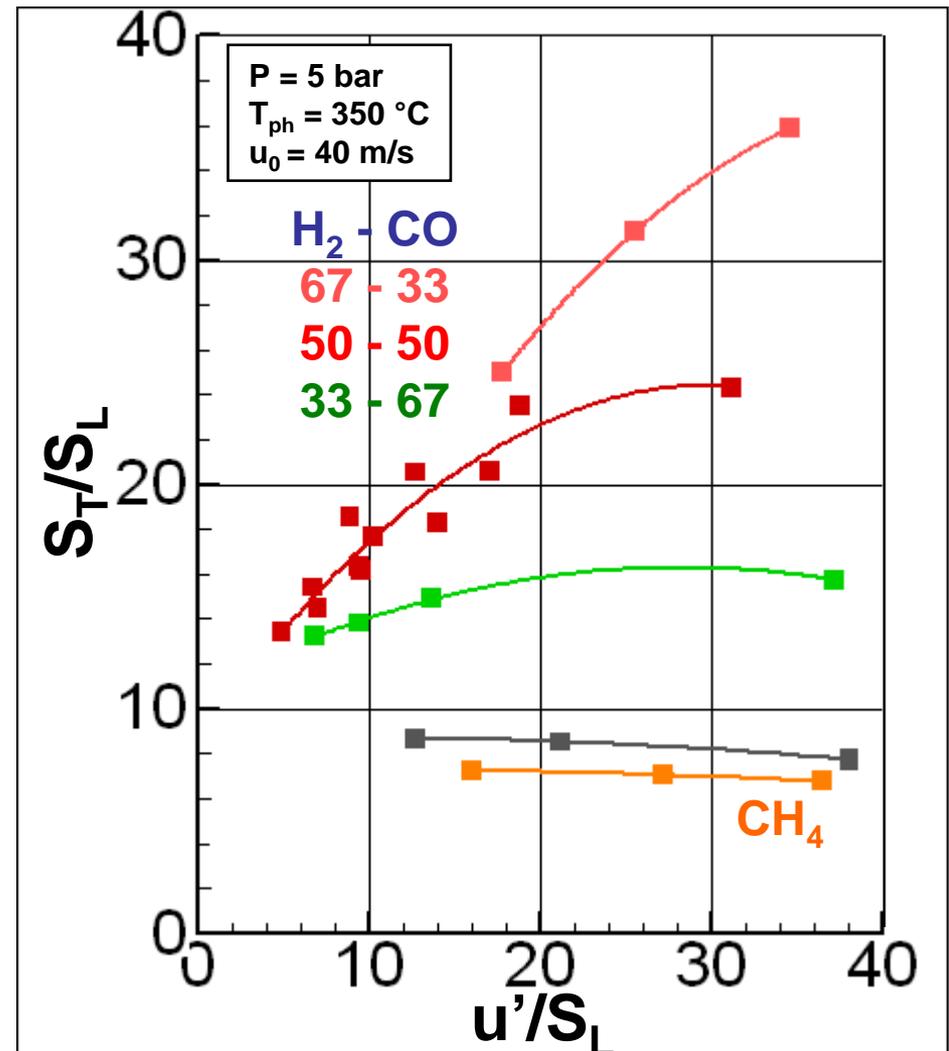
$T_{\text{ph}} = 673$ K

Differences depend on:

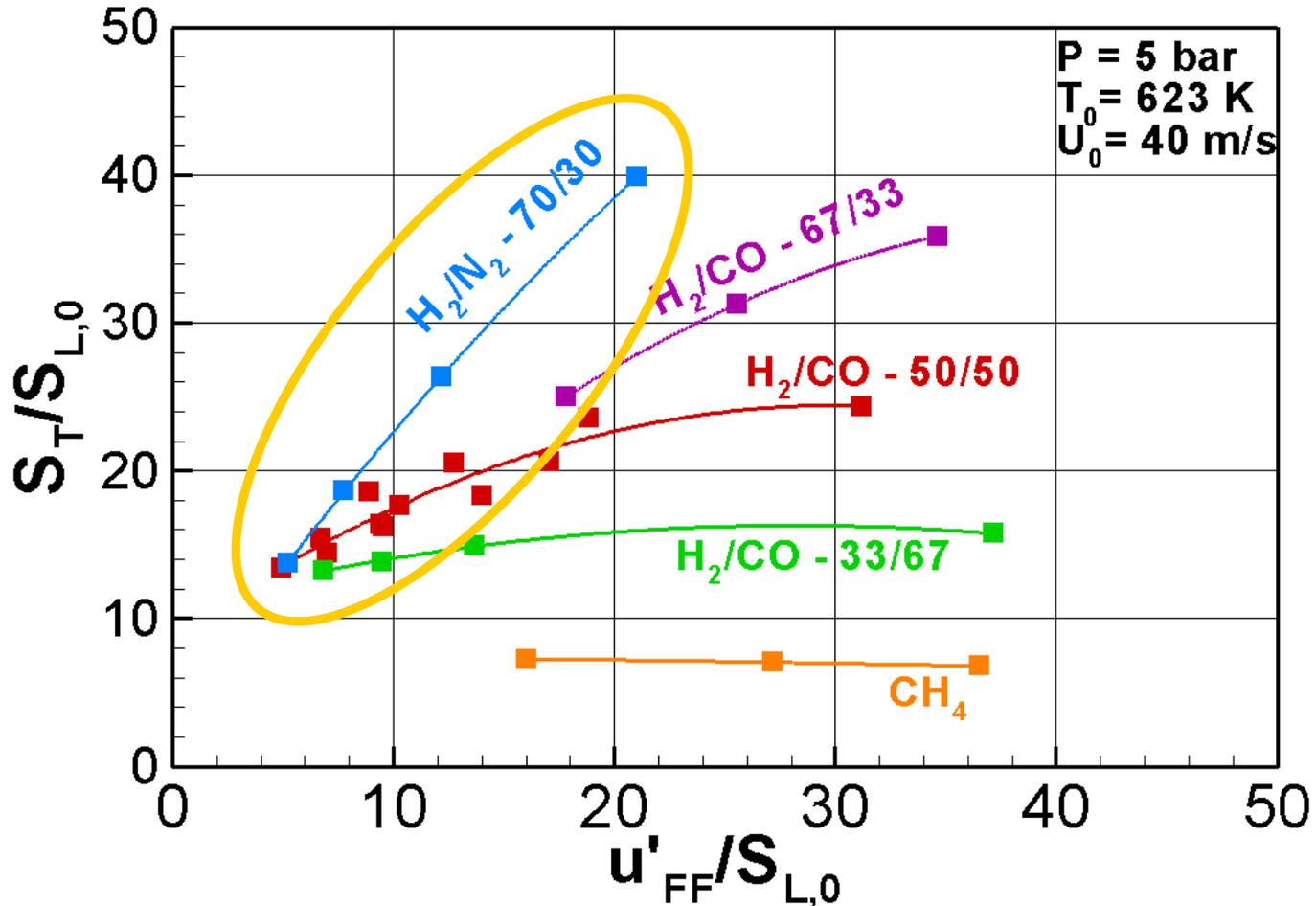
1. Flame temperature
2. Chemistry
3. Turbulence
4. Diffusivity (H2 %)

Normalized Turbulent Flame Speed

- co-firing NG and air-blown SG (@ 60% - 40%) leads to an increase of S_T/S_L of 15%
- firing pure SG increases dramatically the ratio S_T/S_L depending on the H_2 content

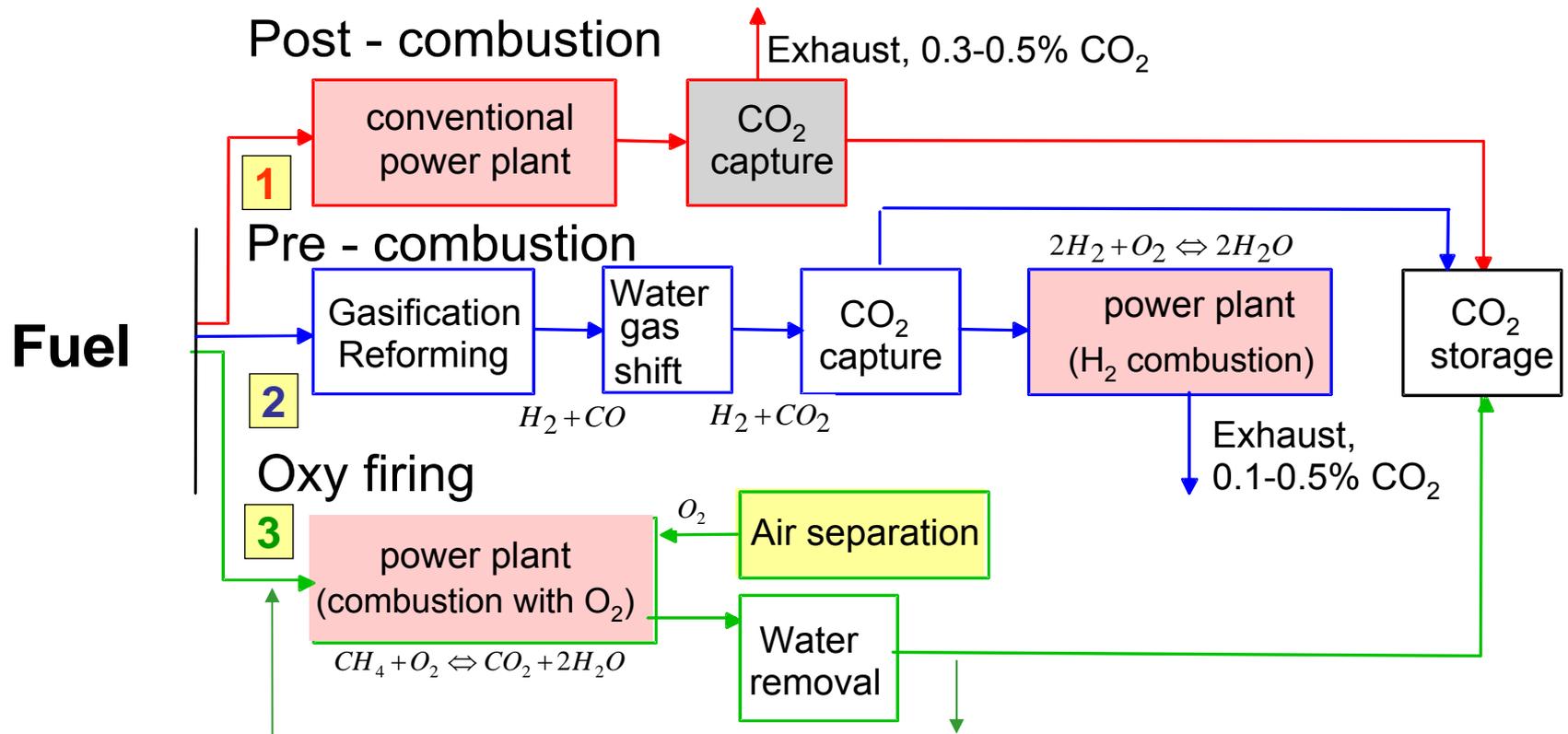


Normalized Turbulent Flame Speed

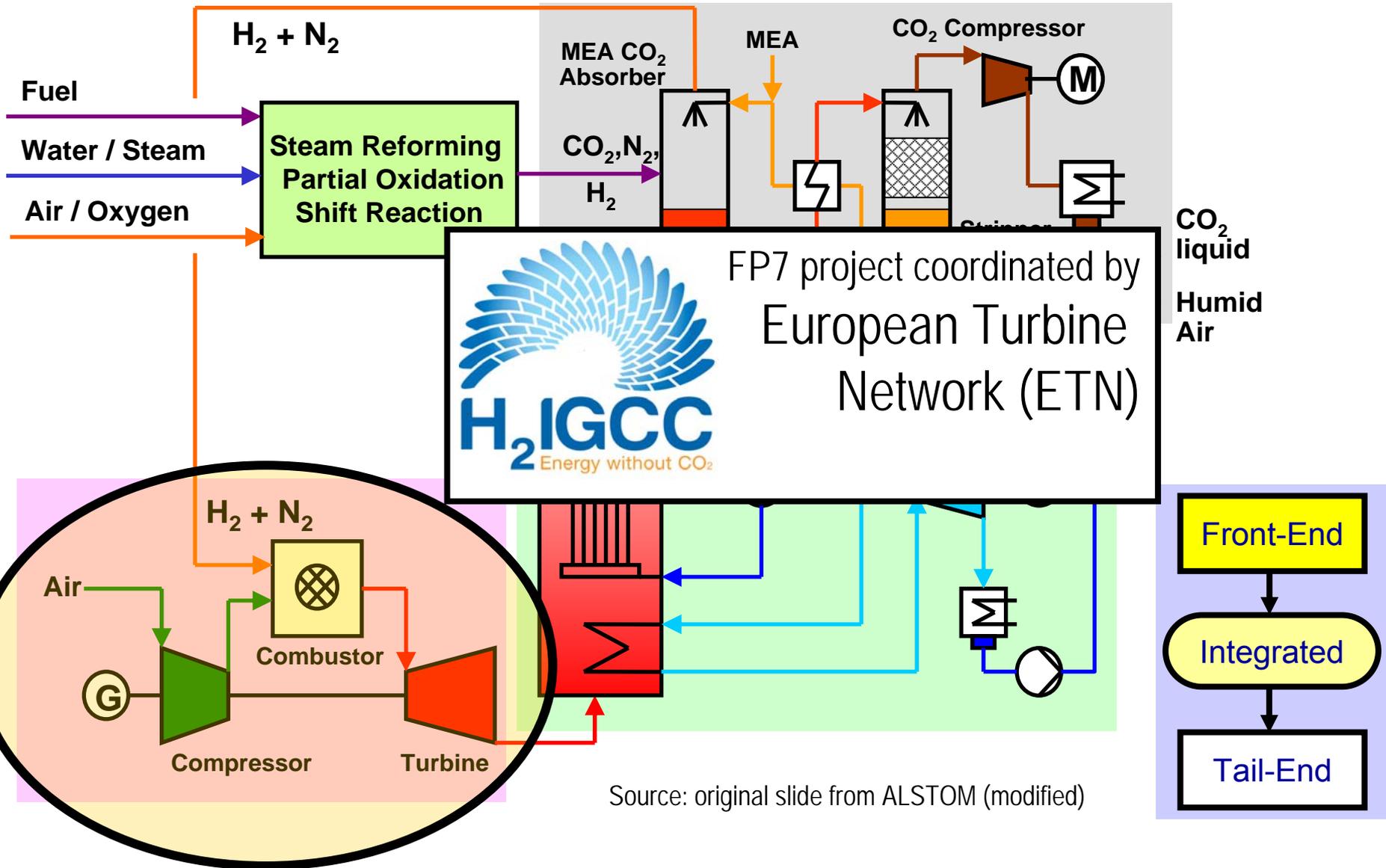


Gasturbine based

Technology options for CO₂-free power generation



Fuel Decarbonization / H₂ combustion





Low Emission Gas Turbine Technology for Hydrogen-rich Syngas

H₂-IGCC Energy without CO₂



Project Details

- Collaborative Project under the EU's Seventh Framework Programme (FP7): Advanced Gas Turbines for Solid Fuel Gasification Processes
- Co-funded by the European Commission, Directorate-General for Energy
- Title: Low Emission Gas Turbine Technology for Hydrogen-rich Syngas
- Acronym: H2-IGCC
- Website: www.h2-igcc.eu
- Duration: 4 years (2009-2013)
- Budget: 17.8 M Euro (11.3 M Euro EU funding)





H₂-IGCC Vision

To pave the way for commercial deployment of efficient, clean, flexible and reliable IGCC-CCS plants by 2020



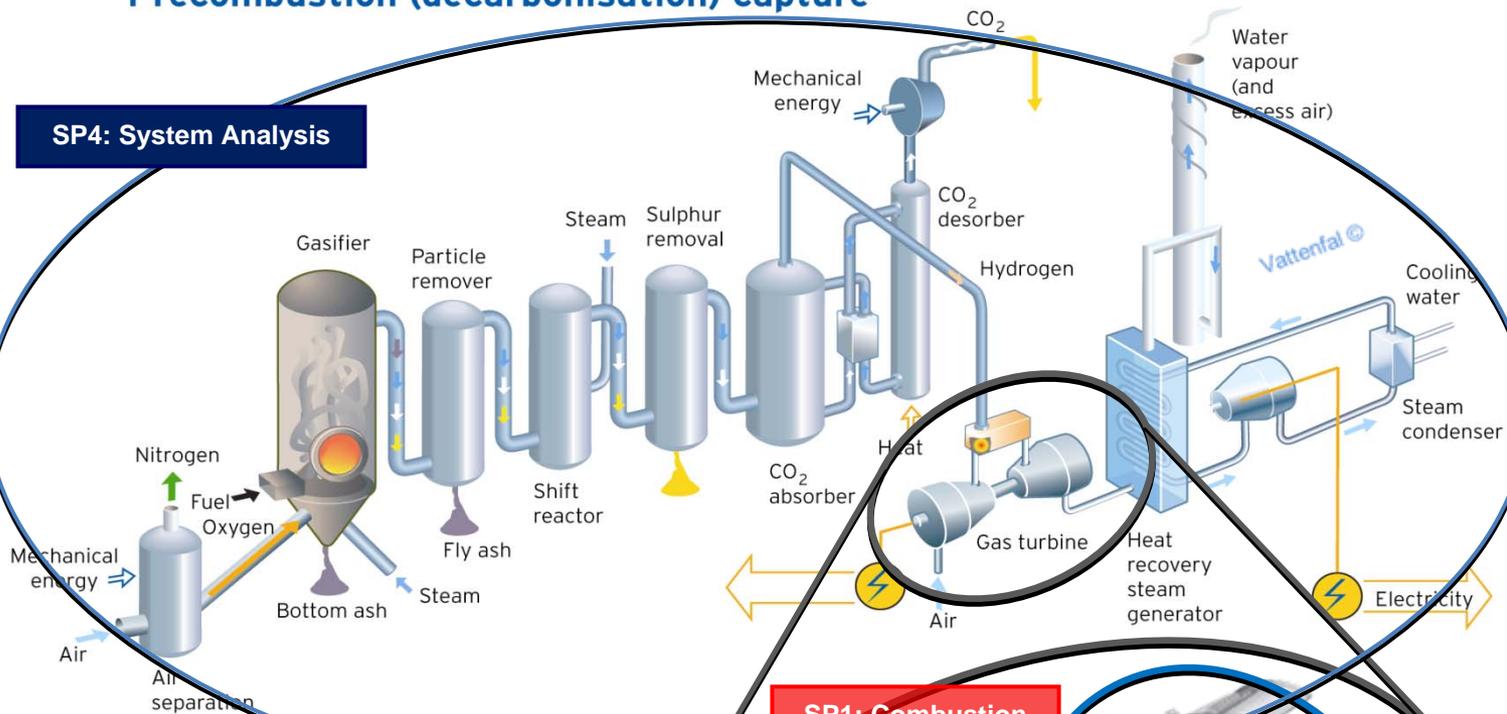
H₂-IGCC project
Co-funded by the European Union
2009-2013

IGCC-CCS Full Scale Demonstrations from 2014

Commercially Available IGCC-CCS Technology by 2020

Energy without CO₂

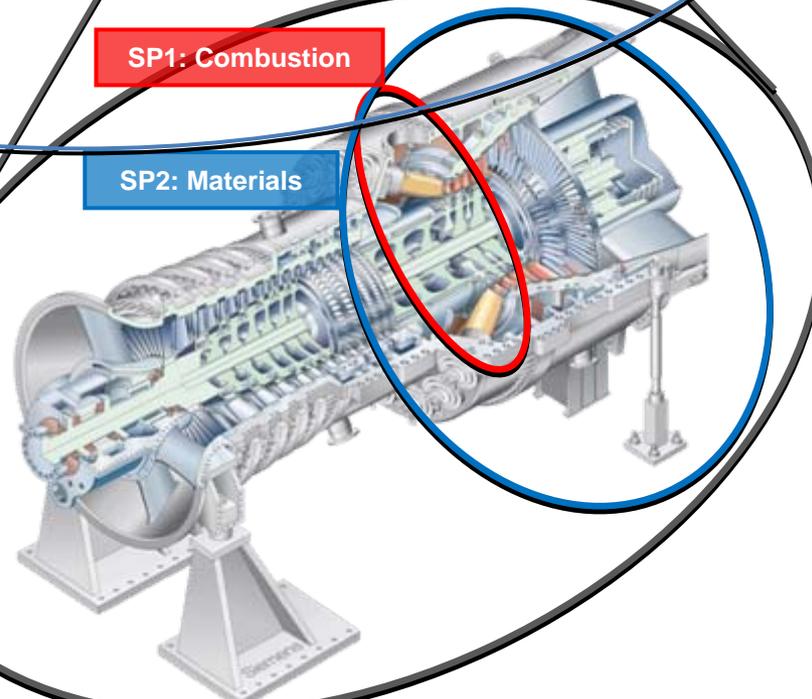
Precombustion (decarbonisation) capture



SP1: Combustion

SP2: Materials

SP3: Turbo machinery



Technical Sub Projects SPs:

SP1: Combustion

SP2: Materials

SP3: Turbo Machinery

SP4: System Analysis



SubProject Partners (11):

Lead

Ansaldo Energia

Utilities

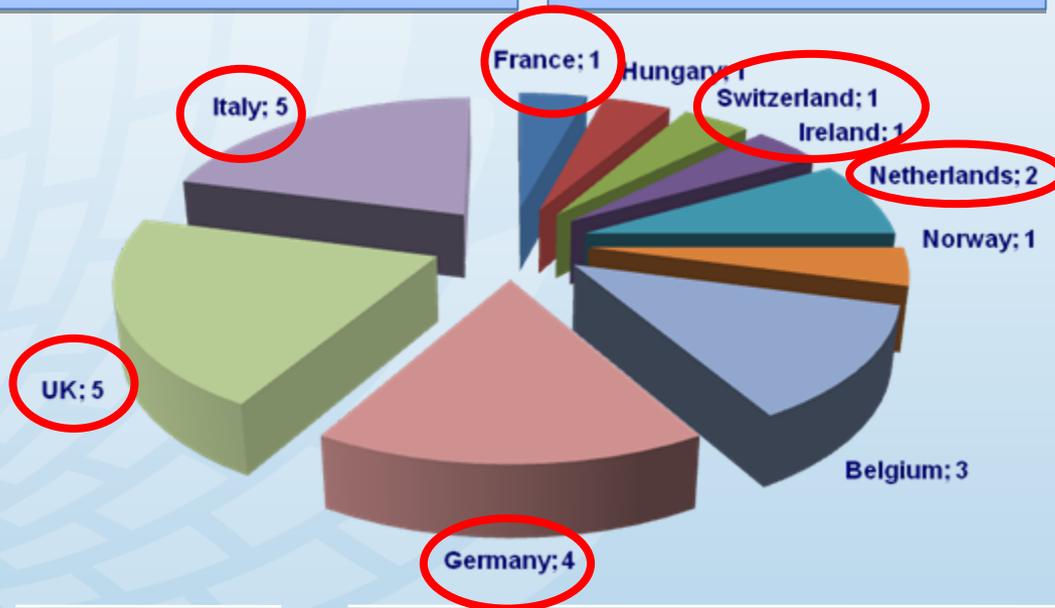
Enel
E.ON Engineering
Electricite De France

OEM's

Ansaldo Energia
Siemens

Research institutes/Universities

DLR
Paul Scherrer Institute
Cardiff University
National Univ. of Ireland Galway
Technical Univ. of Eindhoven
University of Genova



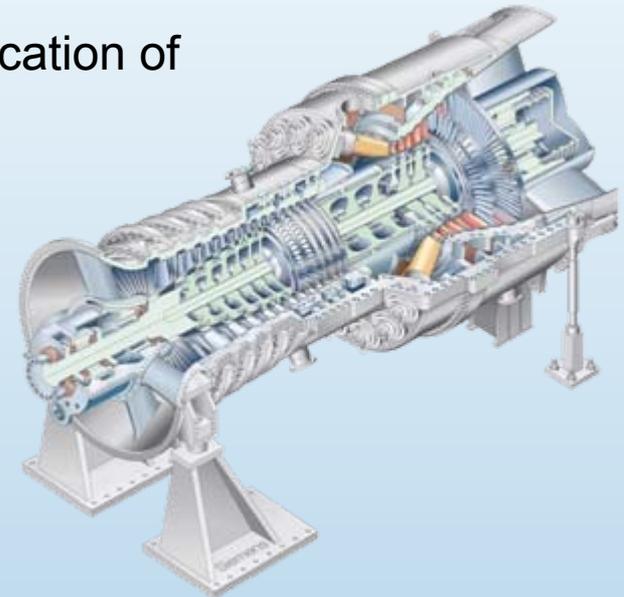
Overall Objective

Safe and low emission combustion technology
for undiluted, hydrogen-rich syngas

- Demonstrate the safe use of undiluted, hydrogen rich syngas in **lean premixed** combustion mode at competitive **low emission levels**
- Demonstrate the safe and competitive use of undiluted, hydrogen rich syngas through the application of **innovative combustion concepts**

Parameter of success

- Demonstrate the use of **undiluted, hydrogen rich syngas** derived from a **pre-combustion CO₂ capture** process in typical **F - class combustion systems** with minimal modifications in order to conserve the **ability to burn a variety of fuels**



Combustion Research @ PSI

*„Contributions to efficient & low emission
energy conversion“*

Peter Jansohn,

Combustion Research Laboratory (CRL)

Results generated during PhD thesis work of

P. Siewert, E. Boschek, S. Daniele

General Energy – Combustion Research Laboratory



Wir schaffen Wissen – heute für morgen